

# Reduction of Ammonia Emission from Stored Laying-hen Manure Using Topically Applied Additives: Zeolite, Al<sup>+</sup>Clear, FERIX-3 and PLT

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**Species:** Poultry (Laying Hens)  
**Use Area:** Manure storage  
**Technology Category:** Chemical Amendments  
**Air Mitigated Pollutants:** Ammonia

## Description:

Ammonia (NH<sub>3</sub>) emissions from animal feeding operations not only reduce the fertilizer nitrogen (N) value of the manure, but also contribute to environmental pollution. Hence, cost-effective means to reduce NH<sub>3</sub> loss associated with animal housing, manure storage and land application will have positive economic and environmental impacts.

Laying hen manure in commercial egg production is typically either accumulated in the lower level of high-rise (HR) houses or removed from manure-belt (MB) cage houses to a manure storage facility one to seven times a week. Various mechanisms are involved in conserving N in poultry manure during storage, such as immobilization of ammonium (NH<sub>4</sub><sup>+</sup>) through addition of easily decomposable, N-poor materials, adsorption of NH<sub>4</sub><sup>+</sup> and NH<sub>3</sub> onto amendments, and regulation of manure pH (Kirchmann et al., 1989).

## Mitigation Mechanism:

Natural zeolite [(Na<sub>4</sub>K<sub>4</sub>) (Al<sub>8</sub>Si<sub>40</sub>) O<sub>96</sub>·24H<sub>2</sub>O] is a cation-exchange compound that has a high affinity and selectivity for NH<sub>4</sub><sup>+</sup> ions due to its crystalline, hydrated properties resulting from its infinite, 3-dimensional structures (Mumpton et al., 1977). It has been used as an amendment to poultry litter, in anaerobic digesters treating cattle manure, in composting of pig slurry and poultry manure, as an air scrubber packing material to improve poultry house environment. Specific research findings include trapping of > 90% of N loss during 13-d composting of pig slurry by placing 12% (by weight) zeolite and chopped straw mixture in the air stream (Bernal et al., 1993); 44% reduction in NH<sub>3</sub> loss during 56-d composting of poultry manure with a surface application of 38% (by weight) zeolite (Kithome et al., 1999); and 22%–47% reduction in NH<sub>3</sub> emissions over 4-d storage of slurry dairy manure when mixed with 2.5%–6.25% (by weight) zeolite (Milan et al., 1999).

Ammonia volatilization stems from microbial decomposition of nitrogenous compounds, principally uric acid, in poultry manure. Manure pH plays a key role in NH<sub>3</sub> volatilization in that NH<sub>3</sub> generation tends to increase with increasing manure pH. Uric acid decomposition is favored under alkaline (pH>7) conditions, and the effect of uricase—the enzyme that catalyzes uric acid breakdown reaches maximum at pH of 9. Consequently, NH<sub>3</sub> emissions can be inhibited by acidulants that lower manure pH and reduce conversion of NH<sub>4</sub> to NH<sub>3</sub>. The acidulants also inhibit bacterial and enzyme activities that are involved in the formation of NH<sub>3</sub>, thus reducing NH<sub>3</sub> production. Liquid Al<sup>+</sup>Clear and dry granular Al<sup>+</sup>Clear (aluminum sulfate), FERIX-3 (ferric sulfate) and PLT (sodium bisulfate) are acidulants that, when hydrated, produce hydrogen ions (H<sup>+</sup>) that attach to NH<sub>3</sub> to form NH<sub>4</sub>. As a result of the reaction, the amount of NH<sub>3</sub> emitting from the manure is reduced, thereby reserving the N content of the manure. Al<sup>+</sup>Clear and PLT had been applied to poultry litter to control NH<sub>3</sub> volatilization (Moore et al., 1995; 1996; Armstrong et al., 2003). FERIX-3 usually is used for industrial and municipal water and wastewater treatment over a wide pH range for color, organics, phosphorous, heavy metal, arsenic and bacteria removal, turbidity, chemical oxygen demand (COD) or biological oxygen demand (BOD) reduction and enhanced coagulation.

## Applicability:

Manure in MB houses drops onto a belt beneath cages and is frequently removed from the house, say, 1 to 7 times per week. The removed manure could be transported and stored in manure storage before it is land-applied or composted. NH<sub>3</sub> emission from manure storage primarily depends on the manure handling practices. The manure surface exposed to the air should be limited to control the NH<sub>3</sub> emission. The following practices are suggested to reduce emissions: 1) reduce the surface area of manure piles; 2) keep adding new manure on the old manure pile; 3) keep the temperature of manure storage low if possible. In addition, using topically applied additive can mitigate the NH<sub>3</sub> emission in the manure storage. This publication covers the testing results of using zeolite, Al<sup>+</sup>Clear, FERIX-3, and PLT as the agents of topical application on nearly fresh laying-hen manure in storage.

## Limitations:

Since Al<sup>+</sup>Clear, FERIX-3 and PLT all have low pH, care must be taken in the material handling (i.e., wearing gloves and eye protection) and to reduce potential corrosion to building components.

## Cost:

Table 1 lists the cost comparison of dry Al<sup>+</sup>Clear, FERIX-3 and PLT, with different application rates. Table 2 lists the costs of the 48.5% liquid Al<sup>+</sup>Clear and Zeolite. Three application rates for liquid and dry Al<sup>+</sup>Clear, FERIX-3, PLT, and Zeolite were tested. Since the NH<sub>3</sub> emission reduction from two of the three application rates for Al<sup>+</sup>Clear, FERIX-3, and PLT had no difference, the costs of higher application rate are not listed for the comparison. The costs listed do not reflect the delivery costs. Because these additives could be applied with the same or similar methods, the comparison does not include the application cost either. The costs of the additives are based on the 50 lb/pack prices of 2008.

Ability of the additives to reduce emissions decreases over time. The costs of the topical application of the agents at end of the 7<sup>th</sup> day was as following: A) 1.56, 1.81 or 1.83 cent/ft<sup>2</sup>-10% NH<sub>3</sub> reduction, respectively, for zeolite applied at 0.6, 1.3, or 1.9 lb/ft<sup>2</sup> (3.1, 6.3, or 12.5 kg m<sup>-2</sup>); B) 0.25 or 0.36 cent/ft<sup>2</sup>-10% NH<sub>3</sub> reduction, respectively, for liquid Al<sup>+</sup>Clear applied at 0.2, or 0.4 lb/ft<sup>2</sup> (1, or 2 kg m<sup>-2</sup>) of manure surface area; C) 0.36 or 0.49 cent/ft<sup>2</sup>-10% NH<sub>3</sub> reduction, respectively, for dry granular Al<sup>+</sup>Clear applied at 0.1 or 0.2 lb/ft<sup>2</sup> (0.5 or 1.0 kg m<sup>-2</sup>); D) 0.46 or 0.42 cent/ft<sup>2</sup>-10% NH<sub>3</sub> reduction, respectively, for FERIX-3 applied at 0.1 or 0.2 lb/ft<sup>2</sup> (0.5 or 1.0 kg m<sup>-2</sup>); and E) 0.45 or 0.60 cent/ft<sup>2</sup>-10% NH<sub>3</sub> reduction, respectively, for PLT applied at 0.1 or 0.2 lb/ft<sup>2</sup> (0.5 or 1.0 kg m<sup>-2</sup>).

**Table 1. Comparison of costs and NH<sub>3</sub> emission reductions (mean ± standard deviation) of topical application of Granular Al<sup>+</sup>Clear, FERIX-3, PLT and Zeolite at different rates on reduction of ammonia emission from stored laying hen manure**

Appl. Rate	kg/m <sup>2</sup> lb/ft <sup>2</sup>	PLT		FERIX-3		Granular Al <sup>+</sup> Clear	
		0.5 0.1	1 0.2	0.5 0.1	1 0.2	0.5 0.1	1 0.2
Cost	Cent per ft <sup>2</sup>	2.5	4.9	1.9	3.8	2.3	4.5
	\$ per m <sup>2</sup>	0.26	0.53	0.21	0.41	0.24	0.48
Cost, cent/ft <sup>2</sup> /10% of NH <sub>3</sub> reduction	Day 1	0.29	0.60	0.22	0.43	0.26	0.50
	Day 2	0.28	0.58	0.22	0.43	0.25	0.50
	Day 3	0.28	0.57	0.22	0.44	0.25	0.50
	Day 4	0.30	0.56	0.22	0.44	0.26	0.50
	Day 5	0.34	0.56	0.25	0.44	0.27	0.50
	Day 6	0.38	0.56	0.32	0.43	0.30	0.49
	Day 7	0.45	0.60	0.46	0.42	0.36	0.49

**Table 2. Costs and NH<sub>3</sub> emission reductions (mean) of topical application of Liquid Al<sup>+</sup>Clear and Zeolite at different rates on reduction of ammonia emission from stored laying hen manure**

Appl. Rate	kg/m <sup>2</sup> lb/ft <sup>2</sup>	Liquid Al <sup>+</sup> Clear			Zeolite		
		1 0.2	2 0.4	3.1 0.6	6.3 1.3	12.5 1.9	
Cost	cent/ft <sup>2</sup>	1.6	3.2	5.6	11.2	16.8	
	\$/m <sup>2</sup>	0.17	0.34	0.6	1.2	1.8	
Cost, cent/ft <sup>2</sup> /10% of NH <sub>3</sub> reduction	1	0.17	0.35	0.85	1.23	1.75	
	2	0.17	0.34	1.04	1.32	1.73	
	3	0.19	0.34	1.19	1.45	1.73	
	4	0.20	0.34	1.30	1.56	1.73	
	5	0.21	0.34	1.40	1.65	1.75	
	6	0.23	0.34	1.47	1.72	1.79	
	7	0.25	0.36	1.56	1.81	1.83	

## Implementation:

Eight emission vessels were designed and built for the study (fig. 1). The vessels were placed in an environmentally-controlled room that was kept at a constant temperature of 23°C (73°F). A flow rate of 3 LPM was introduced into each vessel, resulting in an air exchange rate of 11 air changes per hour (ACH). During each trial, new batch of hen manure was collected and mixed before it was randomly assigned to the eight emission vessels. Manure samples with an initial

weight of 2.5 kg were used as the experimental units. The 2.5 kg sample was placed in a 3.8-liter (1-gal) container (surface area of 0.02 m<sup>2</sup>) that was further placed inside the 19-liter (5-gal) emission vessel.



Figure 1. Emission vessels system used to evaluate efficacy of various manure treatment agents to reduce ammonia emissions from manure storage.

Two vessels were used as controls (i.e., no additives). The five additives tested included natural zeolite, two forms of Al<sup>+</sup>Clear (48.5% liquid and dry granular), FERIX-3, and PLT. The additives were topically applied to the manure samples at three dosages of low, medium, or high that corresponded to 2.5%, 5% or 10% of the manure weight (3.1, 6.3 or 12.5 kg/m<sup>2</sup> manure surface area) for zeolite; 1, 2, or 4 kg/m<sup>2</sup> manure surface area for liquid Al<sup>+</sup>Clear; and 0.5, 1.0, or 1.5 kg/m<sup>2</sup> for granular Al<sup>+</sup>Clear, FERIX-3, and PLT. Two or three trials were conducted to obtain four or six replicates of each treatment. Each trial lasted 7 days.

The NH<sub>3</sub> emission reductions with the medium dosages were in the range of 62% to 93% while NH<sub>3</sub> emission reductions with the low dosages varied from 36% to 82% (Table 3). The NH<sub>3</sub> emission reductions with high dosage were up to 94% during 7-d period. The result shows that there were no significant difference between the high dosage and medium dosage for liquid and granular Al<sup>+</sup>Clear, FERIX-3, and PLT after the 7-d storage period (P=0.5) (Table 3).

Table 3. NH<sub>3</sub> emission reductions (mean ± standard deviation) of topical applied Zeolite, liquid Al<sup>+</sup>Clear, granular Al<sup>+</sup>Clear, FERIX-3 and PLT at different rates at the end of the 7-d storage period

Additives	Application dosage		
	Low	Medium	High
Zeolite	36±10% <sup>a</sup>	62±7% <sup>b</sup>	92±2% <sup>c</sup>
48.5% Liquid Al <sup>+</sup> Clear	63±11% <sup>a</sup>	89±5% <sup>b</sup>	94±2% <sup>b</sup>
Granular Al <sup>+</sup> Clear	81±10% <sup>a</sup>	93±1% <sup>b</sup>	94±2% <sup>b</sup>
FERIX-3	82±5% <sup>a</sup>	86±5% <sup>b</sup>	87±1% <sup>b</sup>
PLT	74±8% <sup>a</sup>	90±3% <sup>b</sup>	92±1% <sup>b</sup>

Table 4. Comparison of NH<sub>3</sub> emission reductions (mean ± standard deviation) of topical applied granular Al<sup>+</sup>Clear, FERIX-3 and PLT at different rates at the end of the 7-d storage period

Appl. Rate	kg/m <sup>2</sup> lb/ft <sup>2</sup>	PLT		FERIX-3		Granular Al <sup>+</sup> Clear	
		0.5 0.1	1 0.2	0.5 0.1	1 0.2	0.5 0.1	1 0.2
Day	1	86±2%	81±3%	87±3%	89±2%	89±1%	91±1%
	2	91±1%	85±2%	88±3%	89±1%	90±1%	91±1%
	3	89±1%	87±3%	88±3%	87±1%	91±1%	89±1%
	4	82±5%	88±3%	86±1%	86±2%	90±1%	89±1%
	5	74±10%	88±2%	77±6%	86±2%	87±2%	90±2%
	6	65±14%	87±2%	60±16%	88±1%	77±5%	91±2%
	7	56±18% <sup>a</sup>	81±2% <sup>b</sup>	42±26% <sup>a</sup>	90±1% <sup>b</sup>	63±8% <sup>a</sup>	92±3% <sup>b</sup>

The granular Al<sup>+</sup>Clear, FERIX-3, and PLT were selected to compare the NH<sub>3</sub> emission reduction rate over a 7-d storage period. In each trial, two vessels were used as controls and two dosages (0.5 and 1.0 kg/m<sup>2</sup>) of Al<sup>+</sup>Clear, FERIX-3, and PLT were applied to the remaining six vessels. During the first 3-d period, there was no difference in NH<sub>3</sub> emission reduction rates among all applications. NH<sub>3</sub> emissions of the low application rate vessels started to increase on 4<sup>th</sup>, 4<sup>th</sup>, and 5<sup>th</sup> d for PLT, FERIX-3, and Al<sup>+</sup>Clear, respectively (fig. 2). At the end of the 7-d period, the NH<sub>3</sub> emission reduction rates from all vessels with 1.0 kg/m<sup>2</sup> application rate were the same (88%), and they were significantly higher than the reduction rates with the 0.5 kg/m<sup>2</sup> rate (Table 4). There was no significant difference in the reduction rates among the three additives with 0.5 kg/m<sup>2</sup> application rate after 7-d period.

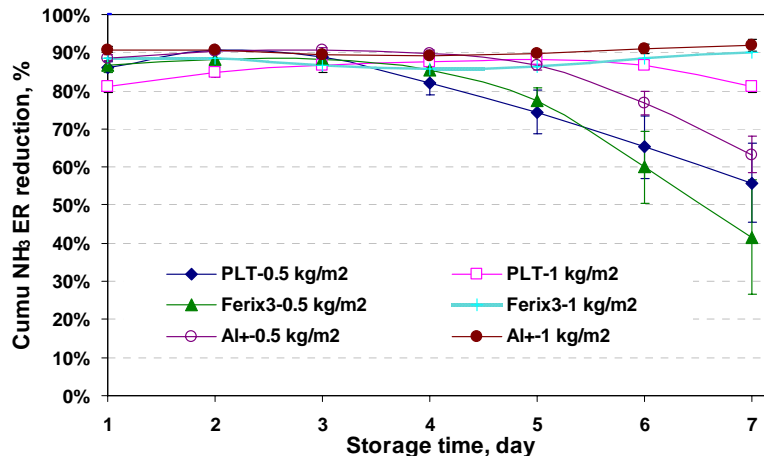


Figure 2. Ammonia emission reductions of ventilated storage of laying hen manure with different rates of topical application of granular Al<sup>+</sup>Clear, FERIX-3, and PLT.

## Technology Summary:

Ammonia emission from manure storage may be controlled by using physical, chemical and/or biological methods. In this study, five treatment agents, including zeolite, 48.5% liquid Al<sup>+</sup>Clear (aluminum sulfate), granular Al<sup>+</sup>Clear (aluminum sulfate), granular FERIX-3 (ferric sulfate), and PLT (sodium bisulfate) were topically applied to stored nearly fresh laying-hen manure. Each agent was tested at three application rates, i.e., low, medium and high. The results show that there were no significant difference between the high and medium dosages for Al+Clear, FERIX-3, and PLT after the 7-d storage period. Reduction of NH<sub>3</sub> emission by the topical application of the agents over a 7-day manure storage/testing period was as following: A) 36%, 62% or 92%, respectively, for zeolite applied at 3.1, 6.3, or 12.5 kg m<sup>-2</sup> (0.6, 1.3, or 1.9 lb/ft<sup>2</sup>); B) 63% or 89%, respectively, for liquid Al<sup>+</sup>Clear applied at 1, or 2 kg m<sup>-2</sup> (0.2, or 0.4 lb/ft<sup>2</sup>) of manure surface area; C) 56% or 81% respectively, for dry granular Al<sup>+</sup>Clear applied at 0.5 or 1.0 kg m<sup>-2</sup> (0.1 or 0.2 lb/ft<sup>2</sup>); D) 42% or 90%, respectively, for FERIX-3 applied at 0.5 or 1.0 kg m<sup>-2</sup> (0.1 or 0.2 lb/ft<sup>2</sup>); and E) 74% or 90%, respectively, for PLT applied at 0.5 or 1.0 kg m<sup>-2</sup> (0.1 or 0.2 lb/ft<sup>2</sup>).

## Additional Resources:

Li, H. 2006. Ammonia emissions from manure-belt laying hen houses and manure storage. PhD diss. Ames, Iowa: Iowa State University.

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## References:

- Kirchmann, H. and E. Witter. 1989. Ammonia volatilization during aerobic and anaerobic manure decomposition. *Plant and Soil* 115: 35-41.
- McCrory, D.F. and P.J. Hobbs. 2001. Additives to reduce ammonia and odor emissions from livestock wastes: a review. *J. Environ. Qual.* 30: 345-355.
- Mumpton, F.A. and P. H. Fishman. 1977. The application of natural zeolites in animal science and aquaculture. *J. Animal Sci.* 45: 1188-1203.
- Maurice, D.V., S.F. Lightsey, E. Hamrick and J. Cox. 1998. Al+Clear sludge and zeolite as components for broiler litter. *J. Appl. Poultry Res.* 7:263-267.

- Bernal, M.P., J.M. Lopez-Real and K.M. Scott. 1993. Application of natural zeolite for the reduction of ammonia emissions during the composting of organic wastes in a laboratory composting simulator. *Bioresource. Tech.* 43:35-39.
- Kithome, M., J.W. Paul and A.A. Bomke. 1999. Reducing nitrogen losses during simulated composting of poultry manure using adsorbents or chemical amendments. *J. Environ. Qual.* 28:194-201.
- Koelliker, J.K., J.R. Miner, M.L. Hellickson and H.S. Nakaue. 1980. A zeolite packed air scrubber to improve poultry house environments. *Trans. ASAE.* 23:157-161
- Milan, Z., E. Sanchez, R. Borja, K. Ilangovan, A Pellon, N. Roviroso, P. Weiland and R. Escobedo. 1999. Deep bed filtration of anaerobic cattle manure effluents with natural zeolite. *J. Environ. Sci. Health. Part B. Pesticides, food contaminants, and agricultural wastes.* B34:305-332.
- Moore, P.A., T.C. Daniel, D.R. Edwards and D.M. Miller. 1995. Effect of chemical amendments on ammonia volatilization from poultry litter. *J. Environ. Qual.* 24:293-300.
- Moore, P.A., T.C. Daniel, D.R. Edwards and D.M. Miller. 1996. Evaluation of chemical amendments to reduce ammonia volatilization from poultry litter. *Poultry Sci.* 75:315-320.
- Lefcourt, A.M. and J.J. Meisinger. 2001. Effect of adding alum or zeolite to dairy slurry on ammonia volatilization and chemical composition. *J. Dairy Sci.* 84: 1814-1821.
- Armstrong, K. A., R. T. Burns, F. R. Walker, L. R. Wilhelm, and D. R. Raman. 2003. Ammonia concentrations in poultry broiler production units treated with liquid alum. Pp. 116-122 in *Air Pollution from Agricultural Operations III, Proceedings of the 12-15 October 2003 Conference (Research Triangle Park, North Carolina USA)*, Publication Date 12 October 2003. 701P1403.

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